

RESPONSE SURFACE MODELING OF WHEAT STRAW PULPING USING SODIUM CARBONATE AND SODIUM HYDROXIDE MIXTURES

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The paper presents the results of experimental trials regarding pulping of wheat straw by using mixtures of sodium carbonate and sodium hydroxide as pulping reagents. Although sodium carbonate has been proved as a potential pulping reagent for obtaining higher yield pulps from wheat straw, the presence of sodium hydroxide improves delignification. Response surface methodology with Central Composite Design has been used as a method for the study of process parameters: temperature, alkali charge and $\text{Na}_2\text{CO}_3/\text{NaOH}$ mass ratio were chosen as independent variables. The studied pulp characteristics were the following: yield, kappa number, tensile index, burst index, corrugating medium flat crush resistance and short span compression resistance. The influence of each variable on pulp characteristics, as well as their interactions, has been established. The results of the study and the obtained quadratic equations model equations may provide the basis for selecting the optimal conditions of pulping according to the desired pulp characteristics.

Keywords: straw, pulping, yield, paper strength

INTRODUCTION

The paper-based products account for about 34% of the packaging market and represent around 50% of the total European paper production. This share is about to increase in the future due to the change in the consumers' preferences towards more environmentally friendly options. The increased demand on paper packaging products at a global scale puts up new pressure on the pulp and paper industry. According to the Confederation of European Paper Industries – CEPI, about 71.7% of used paper products are recovered in Europe (CEPI, 2016). A major part of paper products cannot be recovered and reused because of severe degradation, destruction during usage (*i.e.* incineration), or because of excessive contamination. Other limits of paper recycling include fiber strength loss during paper use and recycling. An important issue regarding an increase in the paper recycling rate lies in the presence of hazardous contaminants from a wide range of sources, including inks, solvents, additives and their degradation products.¹ These

contaminants have led to increased concerns regarding the use of recycled paper products in food packaging in the last years.^{2,3} A recent study indicates the higher content of substances, such as phthalates and phenols, of recycled paper products comparative with virgin fiber ones.⁴

Because of these shortcomings, virgin fibers are needed in the paper industry to replenish the papermaking potential of the recycled fibers, but also for manufacturing of high quality paper products. The virgin fiber share in the fibrous raw materials needed for paper production represents about 44.0%.⁵ Most of the virgin fibers are obtained by wood pulping, which implies a series of environmental impacts, such as deforestation, resource and energy consumption and pollution.

In theory, many vascular plants may be used as raw material for pulping and papermaking. The attractiveness of non-wood fiber sources as raw materials in pulping resides in their lower lignin content and their availability considering their shorter growing cycle.⁶⁻⁸ Most of these resources are abundantly available as waste from cereals or

technical crops harvests in various regions of the world and are either not reused or partially used in other agricultural activities. At present, much of these wastes is returned into soil or incinerated with environmental implications. Among these types of raw materials, wheat straw seems to be the most abundant due to the extended cultivation of wheat. Reports of the Romanian Ministry of Agriculture and Rural Development show that in the year 2015 wheat was cultivated on 2043 thousands of ha, thus being the second most cultivated cereal grain after maize.⁹ In the case of wheat cultures, for 1 tonne unit of wheat about 0.6 to 1 tonnes of straws is generated.¹⁰

There are several factors that prevent the use of straw as raw material for pulping. These factors include: transportation, storage, processing and chemical recovery. These are caused by the low density of straw, their high content of hemicelluloses and silica minerals.^{11,12} Moreover, the selectivity of conventional Kraft pulping is lower than in wood processing, thus economic efficiency is reduced. Therefore, milder reagent pulping processes, such as soda, soda-AQ,¹³ organosolv,¹⁴ aqueous ammonia mixed with caustic potash,¹⁵ sodium carbonate,¹⁶ are recommended. Researchers reported an advantage of non-wood pulp fiber, in particular that obtained from wheat straw, is the improvement of paper resistance to strength loss during recycling.¹⁷

Regarding sodium carbonate usage, in a previous paper,¹⁶ we demonstrated its potential usage as single pulping reagent to produce high yield pulps with strengths suitable for fluting paper. The experimental work presented hereby describes the use of sodium carbonate/sodium hydroxide mixtures as reagents for wheat straw pulping. The aim of the study is to investigate the process factors influence on the pulp characteristics. Temperature, alkali charge and sodium carbonate/sodium hydroxide ratio were chosen as independent variables and their influence on pulp characteristics, as well as their interactions, has been established.

EXPERIMENTAL

Raw materials

Wheat straws – *Triticum aestivum* L., were provided by a Romanian farm. The previously determined chemical composition of the straws included: 43.1% cellulose;¹⁸ 74.4% holocellulose;¹⁹ 27.7% pentosans (TAPPI T 223 cm-10; 17.5% lignin (TAPPI T 222 om-20; 5.5% extractives (TAPPI T 204 om-97 5.3% ash (TAPPI T 211 om-12).¹⁶ The straws

were pre-conditioned by drying up to 8-10% moisture, which was determined according to TAPPI T 664 om-07 standard method.

Pulping of straw and pulp characterization

Pulping trials aiming at obtaining papermaking pulps by alkaline pulping using sodium hydroxide and sodium carbonate mixtures as alkali sources were performed. The pulping trials were performed in a 10 L stainless steel laboratory rotating batch digester, equipped with electric heating and automatic temperature control. Amounts of 300 g (o.d. mass) of straw and a liquor to solid ratio of 5:1 were used in a regular cooking experiment. White liquors were prepared in the laboratory by separately dissolving analytical grade sodium hydroxide (NaOH) and sodium carbonate (Na₂CO₃), in tap water. The heating time was 30 minutes, while the cooking time was set at 60 minutes.

After cooking, the reactor was degassed and cooled to an appropriate temperature to allow removal of pulp for disintegration, washing and squeezing for water removal up to a consistency of about 30%. Furthermore, the yield (gravimetric method) and kappa number (ISO 302:2004) were determined. All specimens were analyzed in triplicates. All chemicals needed for analysis were of analytical grade and purchased from Aldrich.

Paper sheet formation and characterization

The obtained pulp samples were used for laboratory sheet forming after 750 rpm beating on a Jokro mill according to SR EN 25264-3:1997 procedure. Handsheets were obtained according to ISO 5269-2:2004 on a Rapid Köthen laboratory sheet former. The following mechanical properties were determined: tensile strength (ISO 1924:2008); burst strength (ISO 2758:2001); corrugating medium flat crush resistance (also called Concora test CMT-ISO 7263:2011) and SCT – short span compression resistance (ISO 9895:2008).

Experimental design and mathematical modeling

Response surface methodology was the empirical statistical technique employed for the study of the interactions between factors and possible optimization of operating parameters. A central composite factorial design (CCF) was used. The independent variables were: pulping temperature ($X_1 = 150$ °C, 160 °C and 170 °C) alkali charge, as NaOH units ($X_2 = 16\%$, 18% and 20%) and the Na₂CO₃/NaOH mass ratio ($X_3 = 1, 5$ and 9). Each variable was normalized by using Equation 2. The independent variables and their normalized values are presented in Table 1. The start point α value was 1.215. We aimed at establishing quadratic models – Equation 1, by using the experimental data; ANOVA was used to test the significance of the obtained models. The experimental design and data processing were performed with the

help of Modde 11 software. The partial least square fitting method was used to validate experimental data and to generate a mathematical equation that could be used for optimization or forecasting pulping results. The graphical representations of the variations of dependent variables were obtained by Matlab processing of model equations:

$$Y = \beta_0 + \sum_{i=1}^k \beta_i X_i + \sum_{i=1}^k \beta_{ii} X_i^2 + \sum_{i < j} \beta_{ij} X_i X_j + \varepsilon \quad (1)$$

$$X_n = 2 \frac{(X - X_m)}{(X_{\max} - X_{\min})} \quad (2)$$

where X is the absolute (natural) experimental value of the variable concerned; X_m is the mean of the extreme values of X , while X_{\max} and X_{\min} are its maximum and minimum value, respectively.

RESULTS AND DISCUSSION

The experimental results obtained for the pulping trials of wheat straw with mixtures of sodium carbonate and sodium hydroxide are presented in Table 1. The obtained pulp yield values range from 41.84% to 60%, while the Kappa number of the pulps ranges from 35 to 70.4. The obtained yield values are in agreement with those found in other works.^{16,20,21} The Kappa number of pulps is correlated with the yield, in general, with higher values than those obtained in soda pulping, but lower than those obtained in the trials with sodium carbonate used alone as pulping reagent. This can be explained by taking into account that the increases of sodium hydroxide concentration (lower R values) and temperature promote delignification, but also increase polysaccharide degradation, leading to yield losses. Both the yield and Kappa number of pulps are influenced by the process variables and, depending on the severity of the treatment, lower

or higher values may be obtained. The strength properties of the pulps obtained by $\text{Na}_2\text{CO}_3/\text{NaOH}$ pulping also depend on the pulping conditions. The trials characterized by higher severity (*i.e.* high temperature, high alkali charge and low values $\text{Na}_2\text{CO}_3/\text{NaOH}$ ratios) led to lower values of strength properties. The values obtained for the pulp strength properties are comparable with those mentioned in other studies available in the literature.^{20,22}

Response surface methodology may be used to explore the contribution of a given set of independent variables (process parameters or system inputs) on a set of dependent ones (process results or system outputs). Table 1 displays the experimental values that were used for mathematical modeling. Figures 1-6 display the response surfaces and contour plots obtained by using model equations and by maintaining temperature at different constant values.

The coefficient of determination, R^2 , designates the fitness between practical results and the mathematical model. The predictive power of a model is reflected by parameter Q^2 .²³ Modde software computes the R^2 and Q^2 values (displayed in Table 2) and performs the analysis of variance (ANOVA). These coefficient values reflect the validation of prediction. Q^2 is the fraction of the variation of the response predicted by the model according to cross validation and expressed in the same units as R^2 . These parameters are used together as a diagnostic tool for the model. Values close to 1 are desirable, because they indicate an excellent model. In practice, an equal or higher value than 0.5 of Q^2 designates a good model.²³

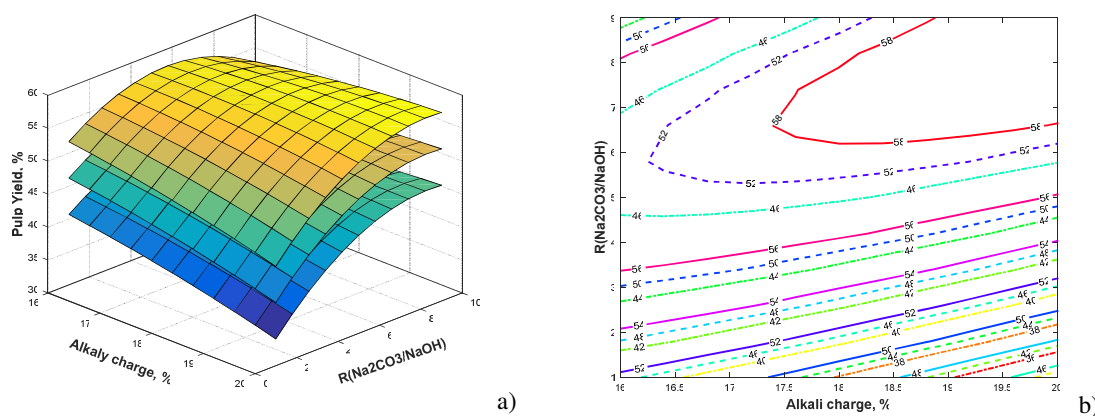


Figure 1: Response surfaces (a) (170 °C – lowest; 160 °C – middle; 150 °C – upper) and contour plot (b) of pulp yield as a function of alkali charge and R (150 °C – continuous line; 160 °C – dotted line; 170 °C – dash-dotted line)

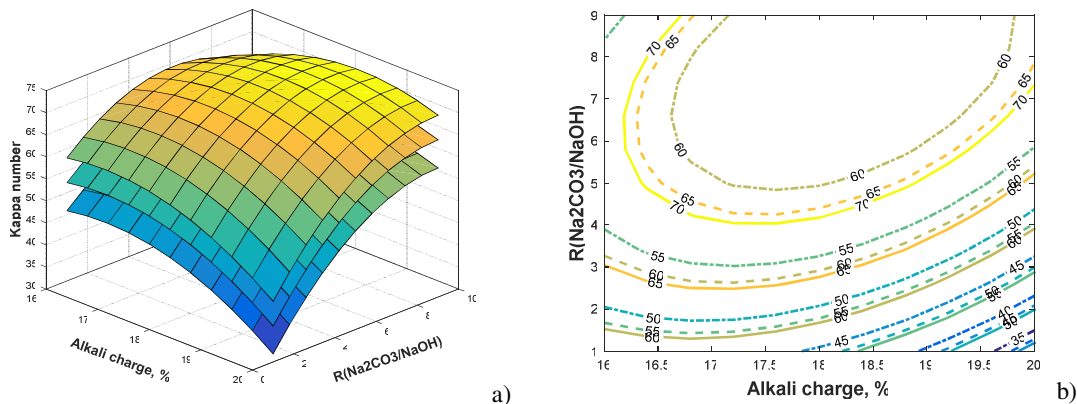


Figure 2: Response surfaces (a) (170 °C – lowest; 160 °C – middle; 150 °C – upper) and contour plot (b) of Kappa number as a function of alkali charge and R (150 °C – continuous line; 160 °C – dotted line; 170 °C – dash-dotted line)

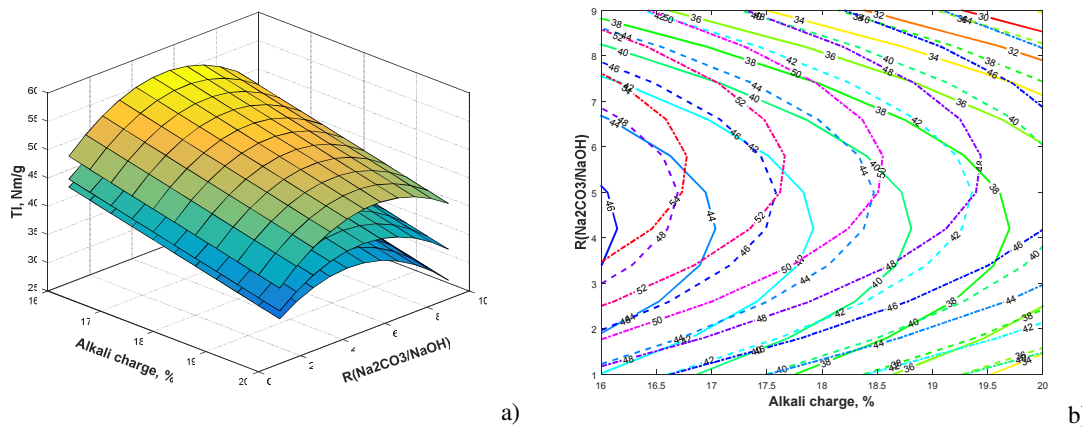


Figure 3: Response surfaces (a) (150 °C – lowest; 160 °C – middle; 170 °C – upper) and contour plot (b) of tensile index as a function of alkali charge and R (150 °C – continuous line; 160 °C – dotted line; 170 °C – dash-dotted line)

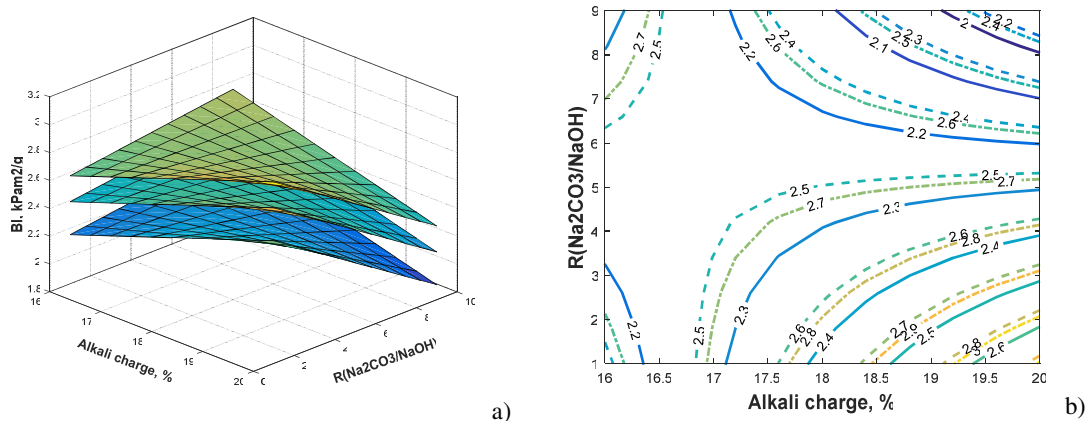


Figure 4: Response surfaces (a) (150 °C – lowest; 160 °C – middle; 170 °C – upper) and contour plot (b) of burst index as a function of alkali charge and R (150 °C – continuous line; 160 °C – dotted line; 170 °C – dash-dotted line)

Table 1
Experimental values used in the response surface methodology study

Run	Normalized variable values X_1, X_2, X_3	Kappa number	Yield (%)	Tensile index (Nm/g)	Burst index (kPam ² /g)	CMT ₀ I (Nm ² /g)	SCTI (Nm/kg)
1	1,1,1	64.2	60.03	44.53	2.12	0.91	19.10
2	1,1,-1	61.5	46.48	24.12	1.91	1.07	23.30
3	1,-1,-1	55.5	47.60	49.17	2.41	1.64	25.90
4	-1,-1, 1	59.0	51.65	40.6	2.00	1.57	25.70
5	1,1,-1	35.5	44.58	36.95	2.51	1.27	25.00
6	-1,1,-1	35.9	51.16	39.96	2.51	1.71	32.20
7	1,-1,-1	41.8	41.91	50.26	2.33	1.56	25.90
8	-1,-1,-1	49.30	50.84	36.50	2.19	1.62	27.20
9	$\alpha, 0, 0$	59.07	45.05	51.24	2.55	1.73	26.50
10	$-\alpha, 0, 0$	66.65	54.18	39.52	2.25	1.66	25.10
11	$0, \alpha, 0$	63.94	52.19	37.29	2.29	1.27	25.00
12	$0, -\alpha, 0$	69.45	50.33	50.64	2.62	1.89	33.10
13	$0, 0, \alpha$	71.40	50.92	33.99	2.36	1.15	21.20
14	$0, 0, -\alpha$	63.90	41.84	33.60	2.56	1.06	21.40
15	$0, 0, 0$	59.40	51.50	47.98	2.47	1.46	28.50
16	$0, 0, 0$	70.40	53.30	46.20	2.45	1.49	26.50
17*	0, 0.5, 0.1	65.15 (66.1)	54.63 (55.7)	46.66 (48.06)	2.58 (2.45)	1.67 (1.55)	22.8 (23.4)
18*	0, 1, 0.17	61.9 (63.99)	46.92 (47.5)	43.91 (46.66)	2.46 (2.40)	1.37 (1.44)	24.5 (25.35)

*trials 17 and 18 were used as control experiments, the observed values are presented together with the predicted values in brackets

Table 2
Equations obtained for the model with coefficients scaled and centered

Dependent variable	Equation	Regression parameters	F	F _{tab.} (p<0.05)
Pulp yield	$Y_{py} = -0.34083X_2^2 - 2.4934X_3^2 - 3.9814X_1 - 0.3196X_2 - 3.5279X_3 + 1.3046X_2X_3 + 51.1783$	$R^2=0.75; Q^2=0.52$	6.15	2.99
Kappa number	$Y_{kn} = -5.7981X_1^2 - 4.039X_2^2 - 4.7062X_3^2 - 3.3334X_1 - 3.2890X_2 + 8.4667X_3 + 3.4553X_2X_3 + 71.006$	$R^2=0.87; Q^2=0.87$	10.44	3.01
Tensile index	$Y_{TI} = 0.8807X_1^2 - 4.81094X_2^2 + 4.2237X_1 - 3.6930X_2 - 0.3744X_3 + 1.6698X_1X_3 + 45.0939$	$R^2=0.83; Q^2=0.52$	7.14	3.37
Burst index	$Y_{BI} = -0.0851X_1^2 - 0.0968X_1 + 0.0152X_2 - 0.1482X_3 - 0.1123X_2X_3 + 2.4819$	$R^2=0.72; Q^2=0.45$	6.69	3.02
CMT ₀ index	$Y_{CMT_0I} = 0.0877X_1^2 - 0.1815X_3^2 - 0.0550X_1X_2 - 0.0915X_2X_3 - 0.0423X_1 - 0.1684X_2 - 0.0744X_3 + 1.5265$	$R^2=0.89; Q^2=0.6$	10.1	3.29
SCT Index	$Y_{SCTI} = 1.0918X_2^2 - 1.122X_3^2 - 0.9557X_1X_2 - 1.1246X_2X_3 - 0.9395X_1 - 1.231X_2 - 1.9343X_3 + 25.7716$	$R^2=0.77; Q^2=0.5$	4.25	3.29

The significance of the model equation for pulp yield, kappa number and mechanical strength properties at 95% confidence interval was checked by the F test, and is shown in Table 2 comparatively with the tabulated F values for probability $p < 0.05$ and corresponding degrees of freedom. The variable importance in the projection values (VIP), computed by Modde, reflects the importance of the terms in the model and the correlation of the terms to all the responses. Equations presented in Table 2 contain the significant model terms with respect to significant VIP score values. The analysis of the data from Figures 1-6 offers information regarding the influence of independent variables on the pulp characteristics.

Figure 1a displays the response surfaces of yield as a function of alkali charge and pulping liquor components mass ratio for different values of temperature. The same may be observed in Figure 1b as a contour plot. As a general rule, the response surfaces and contour plots have been obtained by using the proposed model equations 3-8. A general affirmation is that each pulp characteristic is influenced by a particular set of factors, as revealed in Figure 7, which displays the variable importance in the projection (VIP).

As may be observed from Figure 1 (a and b), the pulp yield depends both on the alkali charge, which if increased, leads to lower yields, and on the ratio of sodium carbonate/sodium hydroxide ratio which if decreased leads to lowering of pulp yields. Pulp yield is inversely varying with the temperature. At constant alkali charge and temperature and with respect to $\text{Na}_2\text{CO}_3/\text{NaOH}$ ratio values, two main variation domains are present – each of them converge to a $\text{Na}_2\text{CO}_3/\text{NaOH}$ ratio around 5 at which maximum yield values are noticed. The variation of pulp yield (Y_{YIELD}) is strongly influenced by temperature, $\text{Na}_2\text{CO}_3/\text{NaOH}$ mass ratio and the interaction of temperature with $\text{Na}_2\text{CO}_3/\text{NaOH}$ ratio. Moderate to low temperature, low alkali charge and high $\text{Na}_2\text{CO}_3/\text{NaOH}$ ratio are needed for obtaining satisfying values of pulp yield.

The response surface and contour plot represented in Figure 2 (a and b) display the variation of Kappa number with respect to alkali charge and $\text{Na}_2\text{CO}_3/\text{NaOH}$ ratio. At constant

temperature and lower alkali charges, high kappa numbers are obtained, while maintaining a constant alkali charge and decreasing $\text{Na}_2\text{CO}_3/\text{NaOH}$ ratio, the pulps' kappa numbers decrease. An increase of temperature also leads to lower values of the kappa number. Similarly to yield variations, at constant alkali charge and temperature and with respect to $\text{Na}_2\text{CO}_3/\text{NaOH}$ ratio values, two main variation domains occur – both of them converge to a $\text{Na}_2\text{CO}_3/\text{NaOH}$ ratio around 5 at which maximum Kappa number values may be observed. As reflected by Figure 7, the pulp Kappa number, Y_{KN} is significantly influenced by terms such as temperature and squared temperature, $\text{Na}_2\text{CO}_3/\text{NaOH}$ mass ratio (and squared) and alkali charge, as well as by the interactions of these terms. The variations of yield and kappa number are explainable by considering the effect of the contribution of all variables to lignin removal, but also to the possible polysaccharides degradation.

As regards pulp strength properties, they are also influenced by the chosen variables, but in a different manner. The tensile index (TI) response surface and contour plot are displayed in Figure 3a and b. The TI of the obtained pulps hits a maximum for lower alkali charges and carbonate/hydroxide ratios. The temperature increase leads to increases of tensile index, but optimal values may be identified at moderate temperatures, moderate alkali charges and $\text{Na}_2\text{CO}_3/\text{NaOH}$ ratio values. These aspects are explainable by the fact that these conditions are sufficient for lignin removal and fiber liberation. The presence of lignin in high yield pulps hinders the formation of inter-fiber bonding, which plays a particular role in the development of mechanical strength.²⁴ On the other hand, at low alkali charges, $\text{Na}_2\text{CO}_3/\text{NaOH}$ ratios and temperature, the pulp lignin content is high and prevents the formation of inter-fibril linkages in paper structures. Figure 7 shows that Y_{TI} is strongly influenced by the first order terms of temperature and alkali charge, second order term of $\text{Na}_2\text{CO}_3/\text{NaOH}$ ratio and, to a minor extent, by interactions between alkali charge and temperature, second order term of alkali charge, temperature, and by first order terms of $\text{Na}_2\text{CO}_3/\text{NaOH}$ ratio.

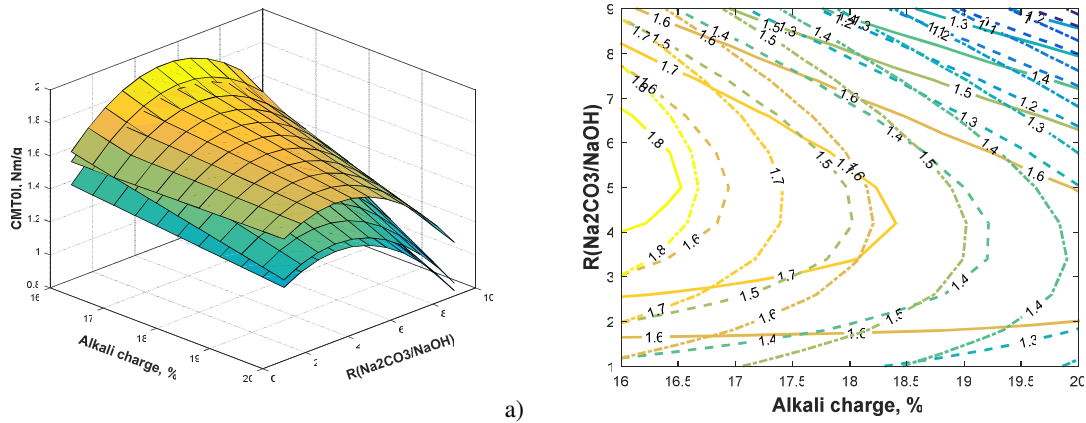


Figure 5: Response surfaces (a) (150 °C – lowest; 160 °C – middle; 170 °C – upper) and contour plot (b) of corrugating medium test index as a function of alkali charge and R (150 °C – continuous line; 160 °C – dotted line; 170 °C – dash-dotted line)

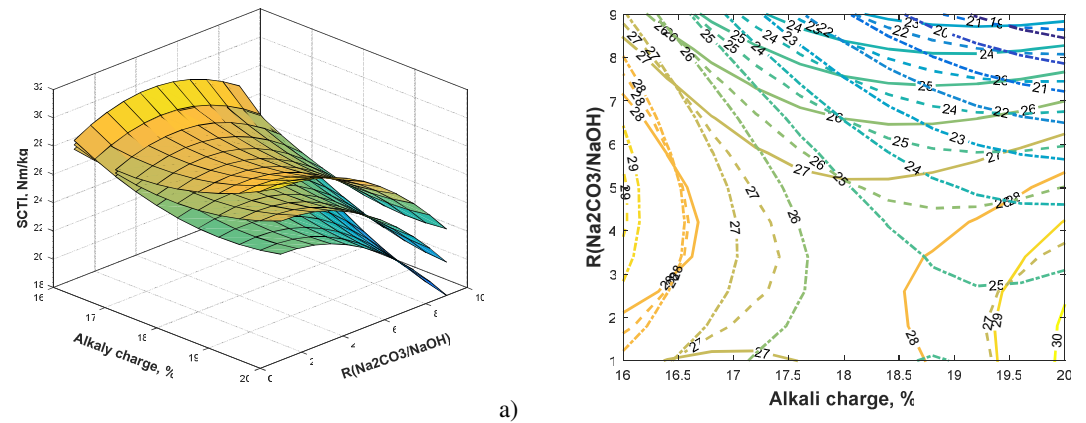


Figure 6: Response surfaces (a) (150 °C – lowest; 160 °C – middle; 170 °C – upper) and contour plot (b) of short span compression test index as a function of alkali charge and R (150 °C – continuous line; 160 °C – dotted line; 170 °C – dash-dotted line)

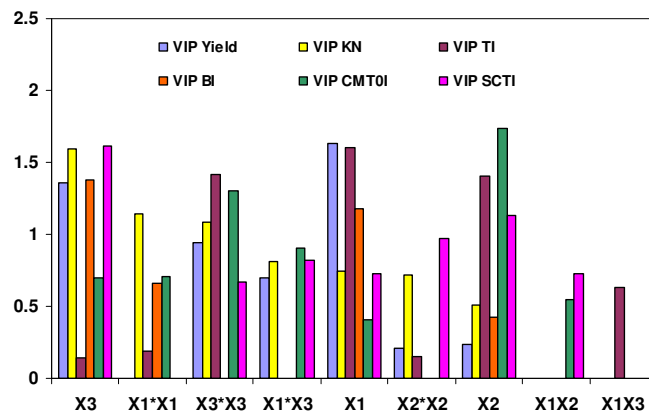


Figure 7: Plot of variables importance projection showing the contributions of independent variables

Table 3
Operational independent variables for optimal values of dependent variables

Response	Optimal values	Independent variable values		
		Temperature (°C)	Alkali charge (%)	Na ₂ CO ₃ /NaOH mass ratio
Yield (%)	58.8	150	20	7.8
Kappa number	31.4	170	20	1
Kappa number and yield	40.9&50.2	150	20	1.1
Tensile index	55.7	169.9	16	5.6
Burst index	2.82	165	20	1
CMT ₀ index	1.9	170	16	5.4
SCT index	30.2	150	20	1

Response surfaces and contour plot of burst index (BI) are displayed in Figure 4a and b. The figure shows higher values of BI for higher pulping temperatures and moderate to high alkali charges and low Na₂CO₃/NaOH ratio. Burst index is strongly influenced by temperature terms (first and second order), Na₂CO₃/NaOH ratio (first order) and to a less extent by alkali charge (first order term) and alkali charge – Na₂CO₃/NaOH ratio interaction.

The SCT index and CMT₀ index are two important fluting paper characteristics, which have been determined by reporting the absolute values of SCT and CMT₀ to the tested paper basis weight value (120 g/m²). The dependence of SCT index on process variables is complex. SCT is positively influenced at lower pulping temperature (*i.e.* 150 °C) and by an increase in alkali charge. The optimal value of alkali charge and carbonate/hydroxide ratio differs as a function of temperature. Independent variables with a high influence on SCT are first order alkali charge, Na₂CO₃/NaOH ratio and temperature; second order alkali charge and Na₂CO₃/NaOH ratio and interactions such as temperature-alkali charge, temperature – Na₂CO₃/NaOH ratio. CMT₀ seems to be less dependent on alkali charge, but shows to decrease at temperature increase. Significant influencing factors on CMT₀ index are: first order alkali charge, temperature and Na₂CO₃/NaOH ratio; second order Na₂CO₃/NaOH ratio and temperature; interactions: temperature-Na₂CO₃/NaOH ratio and temperature-alkali charge.

Optimal conditions for pulping should be chosen according to the targeted pulp characteristics. At constant temperature, if the objective is to obtain higher strength pulps, then the alkali charge should be kept at lower values, while Na₂CO₃/NaOH ratio may take higher values. If lower kappa number pulps are the

target, these values are obtained at lower Na₂CO₃/NaOH ratios and higher alkali charges. In this manner, the pulp producers may choose the pulping conditions that lead to their benefits. Table 3 presents the values of operational independent variables for optimal values of dependent variables. As it may be observed from the data presented in Table 3, there is no particular set of independent variables, leading to optimal values of all the responses.

CONCLUSION

The response surface methodology (RSM) with central composite design is an important instrument for the study and optimization of processes such as pulping, which involve independent variables and response dependent variables. Second order polynomial models were established for describing the influence of pulping temperature, alkali charge and Na₂CO₃/NaOH ratio on the characteristics of the studied pulp. The study of the interaction effects of the independent variables on pulp characteristics was also possible and the influences of each variable on different response hierarchy have been established. The equations provided a good fit relationship between the three pulping variables in general. This would permit the pulp producer to choose the conditions that will lead to the best financial benefits. Depending on the targeted pulp properties and further use, optimal conditions for pulping should be chosen in agreement with the desired pulp characteristics. If the objective is to obtain higher strength pulps, moderate to low temperature pulping should be chosen, with the alkali charge at lower values (16%), while Na₂CO₃/NaOH ratio may take higher values (5-6). If lower kappa number pulps are the target, better values are obtained at lower Na₂CO₃/NaOH ratio values and higher alkali charges, but this occurs with yield losses.

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